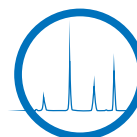


**(U)HPLC  
TROUBLE  
SHOOTING**



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# 1. GENERAL (U)HPLC TROUBLE SHOOTING APPROACH

## **Who hasn't experienced the following?**

Suddenly the chromatographic results have changed.

For example, peak deformation such as splitting or tailing occur, retention time shifts or nothing is detected at all.

Then it is time to find the cause of the anomaly and fix it. Because the final chromatogram is influenced by various factors, all the various components and variables should be examined in a systematically manner.

A systematic review of the following parameters is recommended; each procedure will be described in more detail later in this document:

## System

- **Pumps** (*constant flow? Pressure fluctuation?*)
- **Autosampler** (*Contamination of injection needle? Correct injection?*)
- **Detector** (*UV: Lamp o.k.?*)
- **Column oven** (*Temperature correct?*)
- **Capillaries** (*Correct connection? Fittings correct? Dead volumes?*)

## Eluents

- **Solvents and buffer salts used**  
(*manufacturer changed? Different batch?*)
- **pH value, pH value adjustment**  
(*Changes of pH? pH adjusted? How adjusted?*)
- **Age/Storage** (*Buffer storability? Precipitation? pH change?*)

## Sample

- **Sample changes**  
(*Production changed? Composition? Concentration?*)
- **Sample solvent**  
(*Elution strength? Influence on chromatography?*)

## Column & Guard

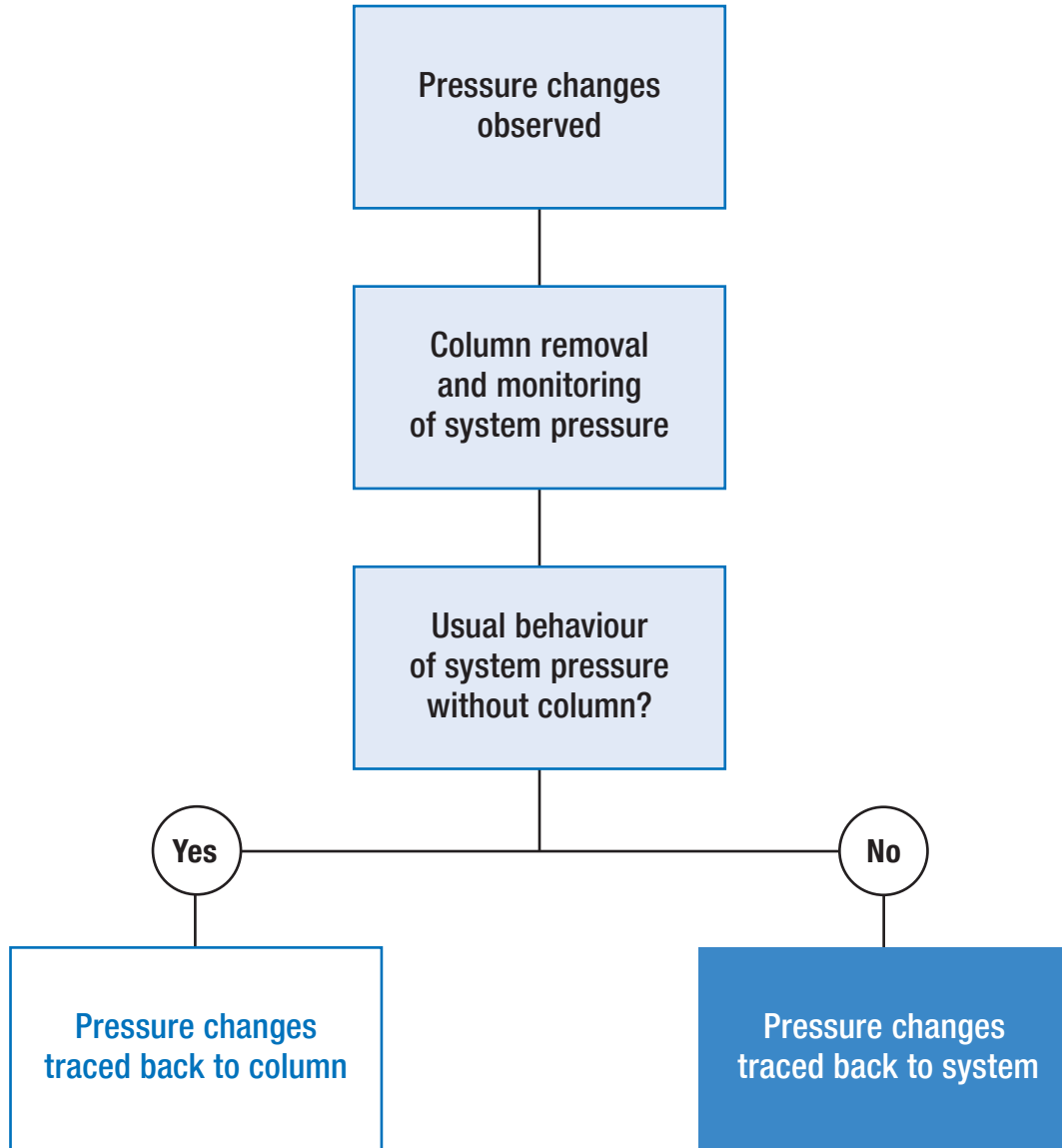
- **Guard column** (*Age? Replacement necessary?*)
- **Column suitability for method** (*pH? Temperature? Eluents?*)
- **Column connection** (*Inlet and outlet correctly connected? Ferrule correctly fitted?*)
- **Column history** (*Injections? Pre-conditioning of the column?*)



## 2. PROBLEMS IN (U)HPLC AND THEIR CAUSE/CORRECTION

### 2.1. Back pressure changes

The root cause of back pressure changes can often be traced back to incorrect handling of samples and solvents which finally results in a blockage or contamination of system components and the analytical column itself.





## No pressure / pressure too low

Potential Causes	Solution
Formation of negative pressure in the eluent reservoir	<ul style="list-style-type: none"><li>• Avoid formation of negative pressure, utilisation of equalising valves in the caps of the eluent flask, prepare fresh buffer</li></ul>
Leakage at piston seal or valves of the pump	<ul style="list-style-type: none"><li>• Check tightness, if necessary, replace seals or valves</li></ul>
Air or particulates in head of pump or valves of the pump	<ul style="list-style-type: none"><li>• Purge the pumps with water or IPA, flush water or IPA with the aid of a syringe through the pump while turned off</li></ul>
Leakage in transmission lines	<ul style="list-style-type: none"><li>• Check and if necessary, exchange screw connections and capillaries</li></ul>

## Increase in pressure / pressure too high

Potential Causes	Solution
Blocked injector or capillaries	<ul style="list-style-type: none"><li>• Clean injector and blocked capillaries or if necessary, replace them</li></ul>
Contamination of water/buffer due to algae / bacteria	<ul style="list-style-type: none"><li>• Prepare fresh buffer (daily)</li></ul>
Blocked guard column or column inlet frit	<ul style="list-style-type: none"><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>
Contamination of stationary phase	<ul style="list-style-type: none"><li>• Wash column with strong solvent</li><li>• Consult the manufacturer's suggested column regeneration procedure</li><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>

## Pressure changes

Potential Causes	Solution
Leakage at piston seal or valves in the pump	<ul style="list-style-type: none"><li>• Check tightness, if necessary, replace seals or valves</li></ul>
Air or particulates in head of pump or valves of the pump	<ul style="list-style-type: none"><li>• Purge the pump with water or IPA, flush water or IPA with the aid of a syringe through the pump when turned off</li></ul>



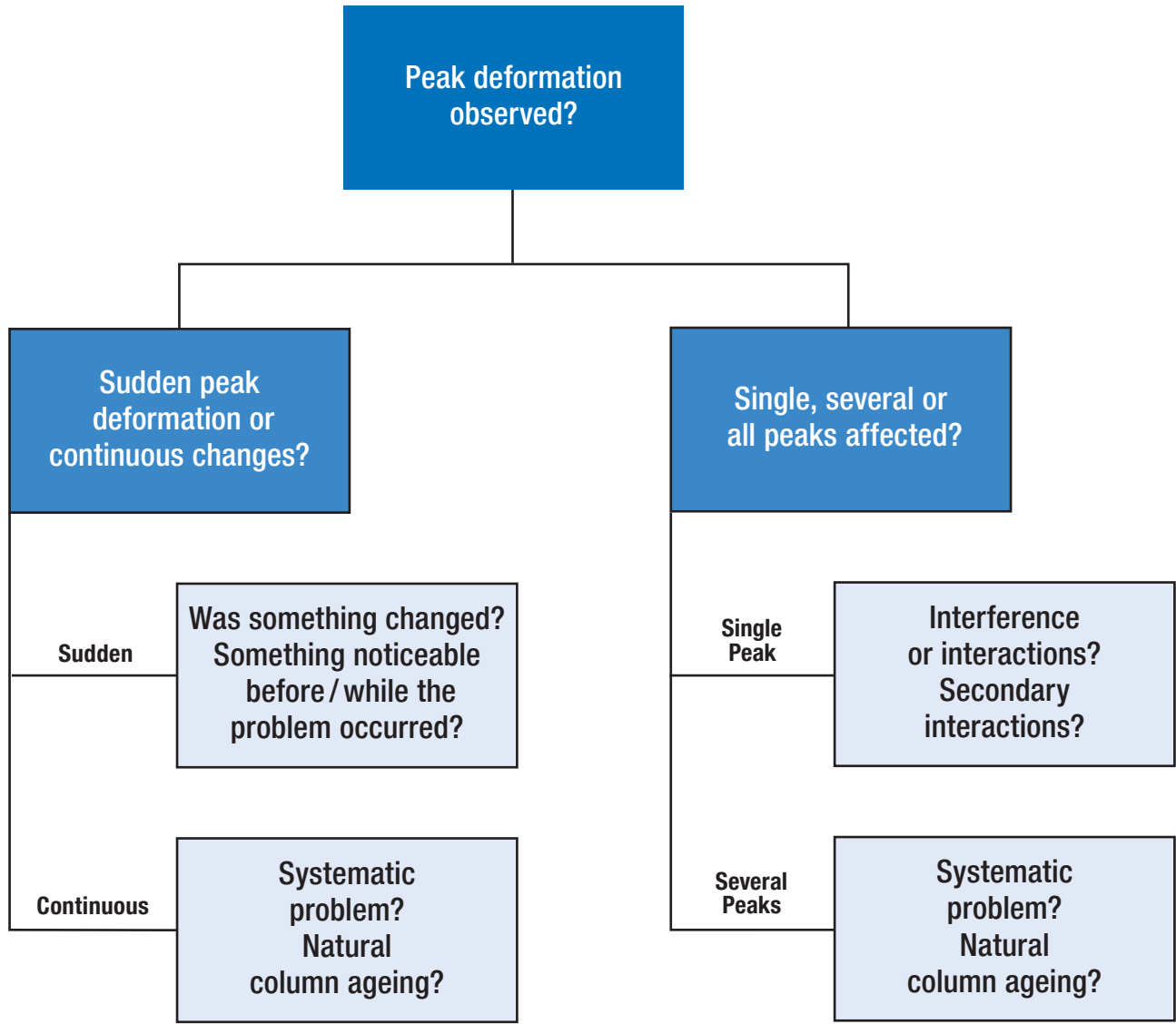


## 2. PROBLEMS IN (U)HPLC AND THEIR CAUSE/CORRECTION

### **2.2. Peak deformation**

In most cases systematic errors in the method conditions, inappropriate choice of column or solvent or natural column ageing are the root causes for peak deformation. By far the most common cause is the use of a too strong or otherwise inappropriate injection solvent, as well as an injection volume that's too large.

Typical symptoms as well as potential causes and solutions are listed in the following tables. These symptoms often appear in combination with other abnormalities in the chromatographic results (e.g., pressure changes or retention time shifts).





# Peak broadening

Potential Causes	Solution
Injection volume too large	<ul style="list-style-type: none"><li>• inject smaller volume</li></ul>
Detection rate too low	<ul style="list-style-type: none"><li>• Increase detection rate</li></ul>
Retention time too long	<ul style="list-style-type: none"><li>• Gradient elution</li><li>• Use of another eluent with higher elution strength</li></ul>
Viscosity of the eluent too high	<ul style="list-style-type: none"><li>• Use a different eluent</li><li>• Increase temperature, use solvent pre-heater</li></ul>
Contamination of the stationary phase	<ul style="list-style-type: none"><li>• Increase column temperature</li><li>• Wash column with strong solvent</li><li>• Consult the manufacturer's suggested column regeneration procedure</li><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>

# Fronting

Potential Causes	Solution
<b>Overloading of column</b>	<ul style="list-style-type: none"><li>• Decrease injection volume or sample concentration</li><li>• Use a column with larger ID</li><li>• Use a stationary phase with higher surface area</li></ul>
<b>Viscosity of sample or mobile phase too high</b>	<ul style="list-style-type: none"><li>• Increase temperature</li><li>• Change to different mobile phase</li><li>• Decrease injection volume or sample concentration</li></ul>
<b>Contamination of the stationary phase</b>	<ul style="list-style-type: none"><li>• Wash column with strong solvent</li><li>• Increase column temperature</li><li>• Consult the manufacturer's suggested column regeneration procedure</li><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>
<b>Needle wash not compatible/too strong</b>	<ul style="list-style-type: none"><li>• reduce solvent strength</li><li>• reduce injector loop volume</li></ul>



# Tailing

Potential Causes	Solution
Interactions of basic analytes with silanol groups	<ul style="list-style-type: none"><li>• Reduce pH value of mobile phase (pH 2–3)</li><li>• Increase ion strength of eluents</li><li>• Switch to a more inert stationary phase</li><li>• Use ion-pairing agents</li></ul>
Dead volume	<ul style="list-style-type: none"><li>• Re-pack column, if possible</li><li>• Replace column</li></ul>
Ageing of packing material due to high temperatures	<ul style="list-style-type: none"><li>• Replace the column</li></ul>
Blocked guard column or column inlet frit	<ul style="list-style-type: none"><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>
Wrong pH value of mobile phase	<ul style="list-style-type: none"><li>• Adjust pH value respectively for stationary phase and analytes</li></ul>
Elution of a second sample component	<ul style="list-style-type: none"><li>• Optimise sample preparation</li><li>• Optimise method parameters</li><li>• Optimise mobile or stationary phase</li><li>• Use a longer column</li></ul>
Contamination of stationary phase	<ul style="list-style-type: none"><li>• Wash column with strong solvent</li><li>• Consult the manufacturer's suggested column regeneration procedure</li><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>

# Split peaks

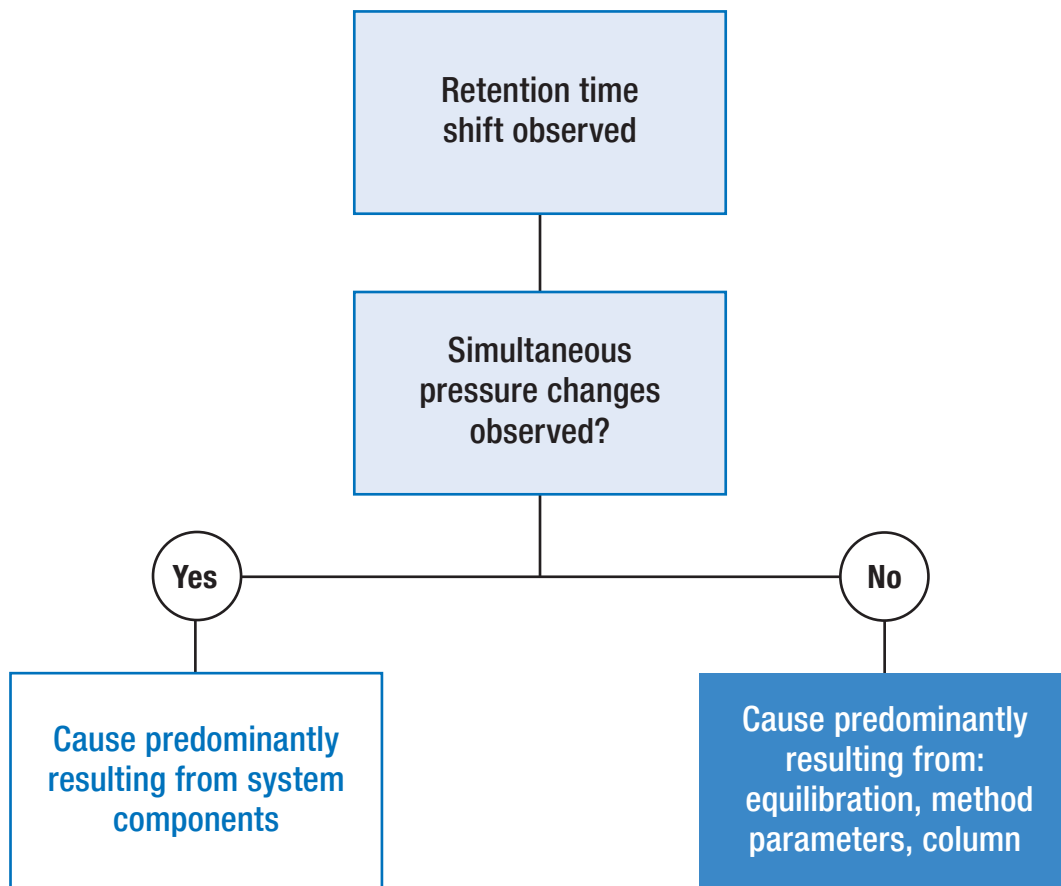
Potential Causes	Solution
Blocked guard column or column inlet frit	<ul style="list-style-type: none"><li>• Backflush the column (if allowed)</li><li>• Replace the column</li></ul>
Overloading of column	<ul style="list-style-type: none"><li>• Reduce injection volume or sample concentration</li><li>• Use a column with larger ID</li></ul>
Inappropriate injection solvent	<ul style="list-style-type: none"><li>• Injection in weaker solvent</li><li>• Adjust pH value of injection solvent to pH of eluents</li><li>• Use a stronger eluent</li></ul>
Dead volume	<ul style="list-style-type: none"><li>• Re-pack column, if possible</li><li>• Replace the column</li></ul>
Elution of a second sample component	<ul style="list-style-type: none"><li>• Optimise sample preparation</li><li>• Optimise method parameters</li><li>• Optimise mobile or stationary phase</li><li>• Utilisation of a longer column</li></ul>
Carry-over of last analysis	<ul style="list-style-type: none"><li>• Clean the column and system</li><li>• Use a more inert stationary phase and column hardware</li></ul>



## 2. PROBLEMS IN (U)HPLC AND THEIR CAUSE/CORRECTION

### **2.3. Retention time and resolution**

A change of retention time and resolution often is accompanied by a change in peak symmetry and efficiency. These are symptoms of a usual column ageing, but can also be caused by leakage in the system, problems with the injector, unstable temperature, or by contamination or damage to the stationary phase due to inappropriate method parameters.







## Retention time is shortened

### Potential Causes

**Overloading of the column**  
(accompanied by too high / broad signal)

**Ageing of the phase /  
phase damaged by harsh conditions**

**Contaminated column**

**Elevated flow rate**

**For phases not stable in 100% aqueous  
conditions: dewetting of the stationary phase**

### Solution

- Reduce injection volume or sample concentration
- Use a column with larger ID
- Use a stationary phase with higher surface area
- Work within the described specifications of the column
- Replace the column
- Prolonged work at the limit of the column specifications will shorten its life time
- Backflush column (if allowed)
- Control flow rate
- Use special aqueous stable phases
- Long equilibration of conventional phases

## Longer retention time

Potential Causes	Solution
Eluent composition has changed	<ul style="list-style-type: none"><li>• Check mixing system</li><li>• Cover storage bottles to avoid a vaporisation of components</li></ul>
Reduced flow rate	<ul style="list-style-type: none"><li>• Control flow rate</li></ul>

## Retention time shift in one direction

Potential Causes	Solution
Leakage of pump seal	<ul style="list-style-type: none"><li>• Replace pump seal</li></ul>
Overloading of the column	<ul style="list-style-type: none"><li>• Reduce injection volume or sample concentration</li></ul>
Insufficient equilibration	<ul style="list-style-type: none"><li>• Equilibration / flush with 10 column volumes</li></ul>



## Retention time fluctuation / random shift

Potential Causes	Solution
Temperature fluctuation	<ul style="list-style-type: none"><li>• Ensure constant temperature</li><li>• For optimum results preheat the mobile phase</li></ul>
Insufficient solvent mixing	<ul style="list-style-type: none"><li>• Evaluate gradient system</li><li>• Where necessary use preheater (mixing chamber)</li></ul>
Wrong buffer concentration or pH value	<ul style="list-style-type: none"><li>• Check eluent pH</li><li>• If necessary, prepare fresh buffer</li></ul>
Insufficient equilibration	<ul style="list-style-type: none"><li>• Equilibration/flush with 10 column volumes</li></ul>
From one system to another: Difference in dwell volume	<ul style="list-style-type: none"><li>• Repeat method with initial system and original conditions</li></ul>
Leakages	<ul style="list-style-type: none"><li>• Search for leakages and seal them</li></ul>
Defective pump valve	<ul style="list-style-type: none"><li>• Check valves and if necessary, exchange them</li></ul>
Air in the pump, air in the mobile phase	<ul style="list-style-type: none"><li>• Degas the mobile phase</li><li>• Inspect system degasser unit</li><li>• If necessary, replace seals</li></ul>
Gradient valve/proportioning valve / MCGV leaks (with low-pressure mixing units, ternary and quaternary systems)	<ul style="list-style-type: none"><li>• flush all mobile phase lines with IPA for 1h</li><li>• if necessary, replace seals</li></ul>

# Loss of resolution

Potential Causes	Solution
<b>Contamination of mobile phase</b>	<ul style="list-style-type: none"><li>• Prepare fresh mobile phase (and replace frequently to avoid contamination)</li></ul>
<b>Blocked pre-column</b>	<ul style="list-style-type: none"><li>• Replace pre-column (in case of change in retention / resolution replace pre-column in time to avoid contamination of main column)</li></ul>
<b>Ageing of stationary phase</b>	<ul style="list-style-type: none"><li>• Flush column regularly and work inside the phase specifications to prevent early ageing of stationary phase.</li><li>• Replace column</li></ul>

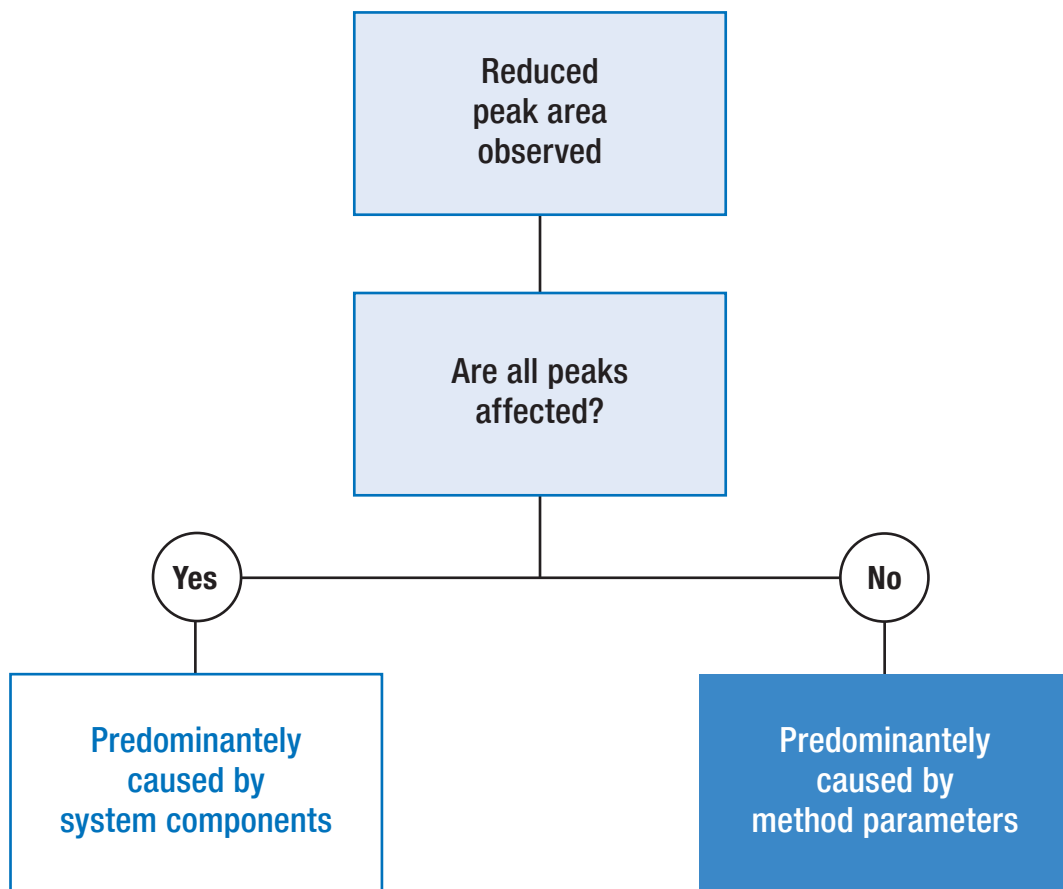


## 2. PROBLEMS IN (U)HPLC AND THEIR CAUSE/CORRECTION

### **2.4. Changes in baseline and peak recovery**

A drifting or noisy baseline can complicate integration of peaks, reduce signal-to-noise ratio and provide decreased sensitivity of the method.

The most common cause for this can be found in the components and settings of the detector used or in a change of eluent composition and can be rectified relatively easily. Further causes might be an inappropriate stationary phase or column hardware, which can lead to signal loss, ghost peaks or carry-over.





## Spikes

Potential Causes	Solution
Air bubbles in mobile phase	<ul style="list-style-type: none"><li>• Degas mobile phase</li><li>• Inspect system degasser unit</li><li>• Verify leak tightness of all connections</li></ul>
Air bubbles in column	<ul style="list-style-type: none"><li>• Always store columns tightly sealed</li></ul>

## Drift to higher signal

Potential Causes	Solution
Build-up and elution of contaminants	<ul style="list-style-type: none"><li>• Optimise sample preparation</li><li>• Use solvents in (U)HPLC grade</li><li>• Follow column cleaning procedure</li></ul>
Mobile phase viscosity is too high	<ul style="list-style-type: none"><li>• Use solvent composition with lower viscosity</li><li>• Increase column temperature</li></ul>
In gradient runs: increasing component B exhibits strong UV-absorption	<ul style="list-style-type: none"><li>• Use solvent with lower or no UV-absorption</li></ul>

## Drift to lower signal

Potential Causes	Solution
In gradient runs: decreasing component A exhibits strong UV-absorption	<ul style="list-style-type: none"><li>• Use solvent with lower or no UV-absorption</li></ul>

## Noise

Potential Causes	Solution
Wave-like noise: Temperature fluctuation in the environment	<ul style="list-style-type: none"><li>• Ensure constant ambient temperature (column oven, isolation of column)</li></ul>
Continuous noise: detector lamp defect or detection cell contaminated	<ul style="list-style-type: none"><li>• Exchange lamp</li><li>• Clean detection cell</li></ul>
Periodic noise: air in pumps	<ul style="list-style-type: none"><li>• Clean pumps</li><li>• Purge air from pumps</li></ul>
Random noise: contamination	<ul style="list-style-type: none"><li>• Optimise sample preparation</li><li>• Clean column</li><li>• Use solvents in (U)HPLC grade</li></ul>
Spikes: air bubbles in detector, eluents or pumps	<ul style="list-style-type: none"><li>• Avoid air bubbles</li><li>• Purge air from components</li></ul>





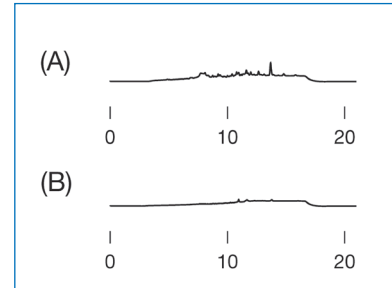
## Negative peaks

Potential Causes	Solution
RI-detector: refractive index of analytes lower than that of mobile phase	<ul style="list-style-type: none"><li>• Switch detector polarity</li></ul>
UV-detector: absorption of mobile phase higher than absorption of analytes	<ul style="list-style-type: none"><li>• Adjust mobile phase</li></ul>

## Reduced peak area

Potential Causes	Solution
Loss of sample due to leakage in injector, capillaries or connections	<ul style="list-style-type: none"><li>• Inspect injector components (valve and needle), connections and capillaries and replace leaky parts</li></ul>
Reduced signal intensity due to contaminated or damaged flow cell/ old UV-lamp	<ul style="list-style-type: none"><li>• Clean flow cell</li><li>• Inspect and replace UV-lamp</li></ul>
Injection volume too low	<ul style="list-style-type: none"><li>• Increase injection volume</li></ul>
High UV-absorption of eluent	<ul style="list-style-type: none"><li>• Use solvents in (U)HPLC grade</li></ul>
Contamination of detector	<ul style="list-style-type: none"><li>• Clean detector</li></ul>
Fluorescence detector: insufficient solvent degassing	<ul style="list-style-type: none"><li>• Clean degasser</li><li>• Reduce flow rate</li></ul>

# Ghost peaks



## Potential Causes

## Solution

**Strong adsorbing component in sample**

- Optimise sample preparation
- Clean column

**Poor quality of water or organic solvent**

- Use (U)HPLC grade water/solvent
- Clean column

**Oxidation of trifluoroacetic acid**

- Prepare eluents daily

**Microbiological growth  
in eluent bottles or system**

- Prepare eluents daily
- Clean the system and column

**Additional analytes in sample**

**Carry-over from last analyses**

- Clean the system and column
- Use a more inert stationary phase and/or inert column hardware

**Air bubbles in mobile phase**

- Degas mobile phase



### 3. COLUMN REGENERATION

Accumulation of contaminants on the stationary phase is a common reason for adverse effects on column performance. This usually manifests itself in:

- **Higher back pressure**
- **Shift in retention time**
- **Deformed peaks**

To avoid accumulation on the column, the use of pre-columns and appropriate sample preparation such as filtration or extraction is advised.

Cleaning procedures are an effective measure for column regeneration in case of adsorption which has already occurred on the material of the stationary phase or to prevent accumulation of contaminants in the first place. Because contaminants usually accumulate and remain at the column head, cleaning procedures should always flush the column in the opposite direction to the usual flow to remove the contaminants more easily and not spread them throughout the whole stationary phase. Sufficient cleaning can be accomplished with a minimum of 20 column volumes of appropriate solvent.

When choosing the appropriate solvent, properties of the contaminants as well as stability of the column used should be considered. Examples for suitable solvents depending on the contamination present are listed in the following table. Higher efficiency of the cleaning procedure can be achieved by increasing the column temperature. Depending on the solvent and the stability of the stationary phase 40°C – 90°C is an appropriate temperature range for this.

For further recommendations follow the instructions of the column manufacturer, usually found in the **“Care and Use Instructions”**.

# Removal of contamination due to different substances

## Salts

- Water and aqueous organic solvents

## Non-polar substances

- Acetonitrile
- Isopropanol
- Tetrahydrofuran
- Dichloromethane
- Chloroform

## Polar substances

- Water and aqueous organic solvents
- Methanol
- Tetrahydrofuran

## Proteins

- Injection of dimethyl sulfoxide
- Gradient starting at 10% to 90% B with A = 0.1% TFA in water and B = 0.1% TFA in acetonitrile



## 4. PREVENTATIVE MEASURES

General considerations:

- **Frequent cleaning of system, column and injector**
- **Regular replacement of worn and damaged parts such as filters, frits and seals**

to avoid contamination.

Additionally, troubleshooting can be simplified immensely by frequently recording a column journal. We recommend you record the identity of the column, number of injections, resulting backpressure, analytes and method used and regularly verify the quality of the column with the results of a standard test method. Switching of eluent bottles, error messages or other abnormalities should also be recorded.

A column journal can reduce the duration of troubleshooting from days to minutes and can additionally provide a way to track your productivity gain after the optimization of your method parameters.

## Use of guard columns

A guard column or guard cartridge systems protects the main column from a build-up of contaminants by absorbing them before they can reach the column.

Therefore, guards prevent:

- **Fast and early ageing of the column**
- **Shift of retention times**
- **Peak deformation**
- **Void formation in stationary phase due to pressure pulses**

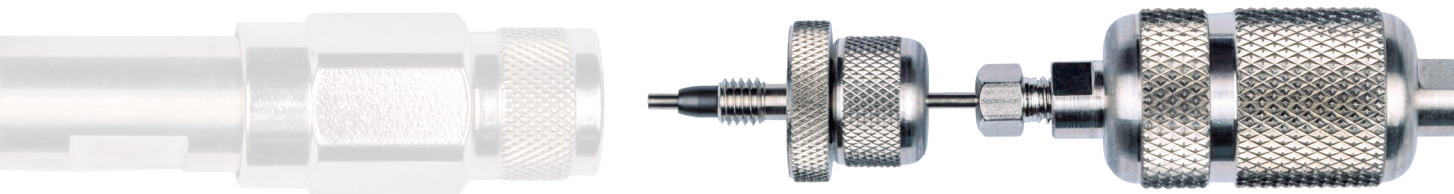


FIGURE 1: UNIVERSAL GUARD COLUMN HOLDER WITH GUARD CARTRIDGE.

Guards should be replaced regularly e.g., when back pressure increases or retention times shift. With a general test method, you can record resolution, peak symmetry and back pressure under known conditions to evaluate the status of your column and guard.



## Zero dead volume connections

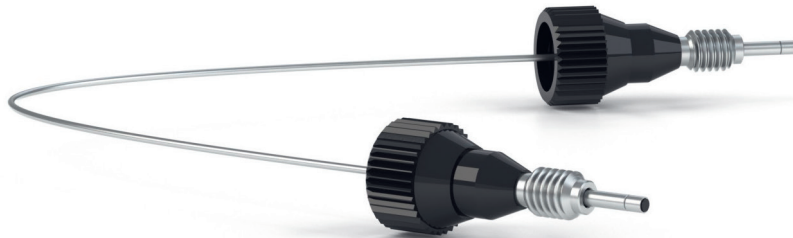


FIGURE 2: MARVELXACT ZERO DEAD VOLUME UNIVERSAL (U)HPLC CONNECTOR.

Especially in UHPLC applications additional column dead volume leads to band broadening. To prevent broad peaks due to non-optimal connections the use of zero dead volume universal connectors such as MarvelXACT is recommended. Additionally, all capillaries in the chromatographic system should be as short as possible.

## Filtration of sample and eluents

Before use the sample and eluents should be filtered with a filter pore size of  $0.45\ \mu\text{m}$  (for  $> 3\ \mu\text{m}$  particle phases) or  $0.2\ \mu\text{m}$  (for  $\leq 3\ \mu\text{m}$  particle phases) depending of particle size of the stationary phase.

Protection against:

- **Blockage of the inlet frit**
- **Particulate contamination of the stationary phase**
- **Peak splitting**
- **Fronting**
- **Tailing**
- **Increased back pressure**

## Frequent replacement of eluents

Due to ageing of the eluents (by biological growth in aqueous buffers, polymerisation of pure acetonitrile, etc) the chromatographic system can become contaminated.

Replacement of aqueous mobile phase is recommended:

**HPLC:** at least once a week

**UHPLC:** at least every three days

## Degassing of eluents

Eluents can dissolve air and contain air bubbles, which can lead to variations in retention times.

Always degas all eluents after preparation and filtration by:

- Purging the eluent with helium (daily)
- Use online vacuum degassing

## Adjust sample solvent to mobile phase conditions

The sample solvent can have a strong influence on the quality of the separation and a suitable composition should be considered:

- Use a weak solvent composition for the sample → **or fronting of early eluting peaks**
- Avoid large differences in the pH of sample and eluents → **or peak splitting**





## Sufficient equilibration

When equilibrating and cleaning a column there's usually the question:

How long does the column have to be flushed?

If the equilibration time is too long you lose productivity and increase cost. If it is too short, there is a risk of non-reproducible results i.e., because of retention time shifts. So in case of doubt, to ensure correct results a longer duration is more favourable.

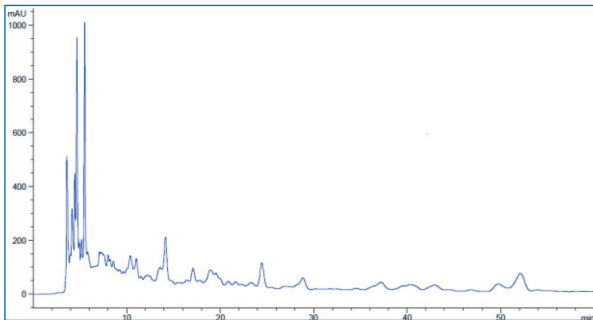
### Example:

Different test conditions were chosen for the analysis of a plant extract to optimise the duration of the equilibration of a YMC-Triart C8 column (column volume 2.5 mL): a usual result (5 CV equilibration), without equilibration, equilibration with 1 column volume, equilibration with 10 column volumes.

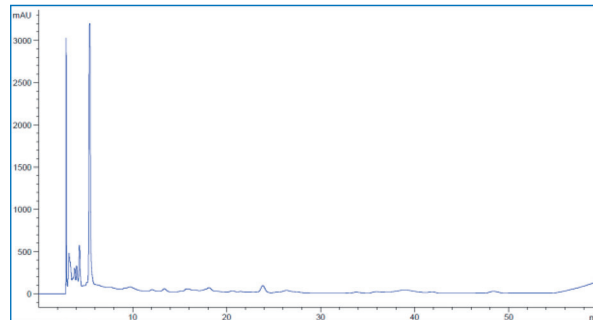
The column was flushed with 50 mL (= 20 CV) of 100 % acetonitrile after each test cycle.

The equilibration and isocratic separation were achieved in a water/acetonitrile (80/20) mixture.

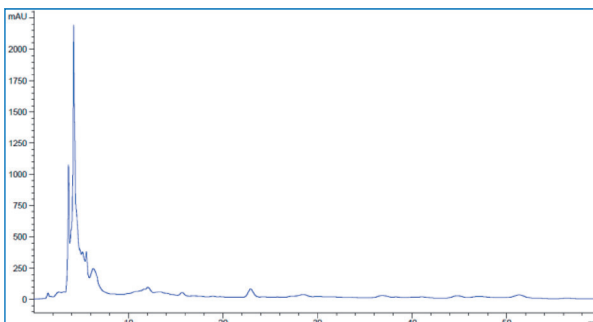
The dwell volume of the system was 130  $\mu$ L.



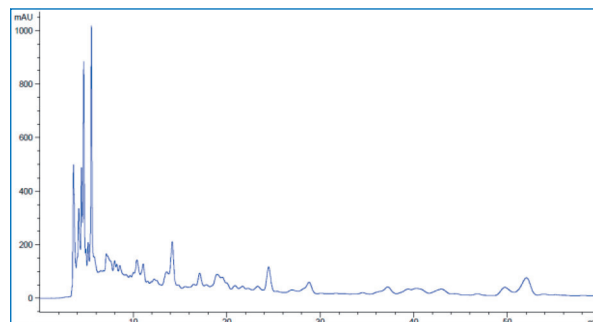
1. Usual result (5 CV).



2. Without equilibration.



3. After insufficient equilibration (1 CV).



4. After sufficient equilibration (10 CV).

It is shown that without sufficient equilibration, retention times shorten significantly, whereas equilibration with 5 or 10 CV leads to a reproducible application.



## Estimation of equilibration duration

How can the optimal equilibration duration be estimated?

→ by means of the geometrical column volume!

The geometrical column volume is a useful parameter to estimate the necessary solvent volume for cleaning and equilibration steps in HPLC – a requirement for reproducible and valid results.

$$\text{Geometric column volume [mL]} = \text{length [cm]} \times (\text{radius [cm]})^2 \times \pi$$

### Calculation of column volume

*Example:*

column:	<b>YMC-Triart C18</b>
column dimensions:	250 x 4.6 mm ID
column volume[mL]	= 25 cm × (0,23 cm) <sup>2</sup> × 3,14
	= 4.2 cm <sup>3</sup>

Table 1: overview of geometrical column volumes [mL] for selected column dimensions

∅ [mm] \ L [mm]	50	75	100	150	250	300
2.0	0.2	0.2	0.3	0.5	0.8	0.9
3.0	0.4	0.5	0.7	1.1	1.8	2.1
4.6	0.8	1.2	1.7	2.5	4.2	5.0
6.0	1.4	2.1	2.8	4.2	7.1	8.5
8.0	2.5	3.8	5.0	7.5	12.6	15.1
10.0	3.9	5.9	7.9	11.8	19.6	23.6
20.0	15.7	23.6	31.4	47.1	78.5	94.2
30.0	35.3	53.0	70.7	106.0	176.7	212.1
50.0	98.2	147.3	196.3	294.5	490.9	589.0

## Additional considerations

### Dwell Volume

- Volume of the system before solvent reaches the column.
- Dependant on system (ask the manufacturer)

### Volume of the stationary phase

- Include correction factor (0.6–0.8)
- Dependant of stationary phase and packing density



# Advanced training and seminars

For several years YMC has combined theoretical and practical know-how in all areas of liquid chromatography.

## Use our seminar offer to expand your skills!

We will gladly offer you a custom seminar package and workshop for your specific topics and scope of work. Amongst other topics you can choose and select from the following subjects:

- **HPLC troubleshooting**
- **HPLC basics**
- **Stationary phase selection**
- **Method development**

### Your benefit:

Due to small group sizes and presentation by two speakers you can expect interactive courses in an open and friendly atmosphere. Your questions and real-life subjects can be specifically addressed and discussed individually.

Further topics are targeted on practical and current chromatographic subjects.

### Choose your seminar:

- **Preparative chromatography- method development and scale-up**
- **Glass column packing- laboratory- and pilot-scale**
- **Method development in chiral chromatography**

Seminar hand-outs, a certificate of attendance and catering are included of course!

# In-house and online seminars – you choose the location!

We gladly welcome you in to our seminar rooms, online or alternatively we can visit you at a location of your choosing!

## **Useful if:**

You can't easily come to one of our seminars?

You want some of your colleagues to benefit from our seminar offer as well?

You expect a custom workshop that specifically addresses all of your needs and work topics?

You can book all of our workshops as in-house seminars or online courses. Just choose from the seminar topics we suggest or define different subjects you're interested in yourself.

Date, location and number of participants can be flexibly coordinated with us.

We would be glad to discuss options in a personal consultation!

## **Your benefits:**

- **Custom seminar topics and flexible scope of the courses**
- **Location and date for the seminar by arrangement**
- **Flexible number of attendees by arrangement**
- **Fixed price**

What can we do to help you? Please approach us for your individual consultation.

**Dr. Mathias Hehn**

**Head of Laboratories**

Phone +49 2064 280

Fax +49 2064 427-222

Email [info@ymc.eu](mailto:info@ymc.eu)

[www.ymc.eu](http://www.ymc.eu)



# Direct Support

If you have an urgent issue and need immediate support, our team of experts is at your disposal!

Contact us via email ([support@ymc.eu](mailto:support@ymc.eu)) or phone!

**Dr. Friederike Becker**  
**Director Marketing and Technology**

Phone +49 2064 427-280

Fax +49 2064 427-222

Email [f.becker@ymc.eu](mailto:f.becker@ymc.eu)

**Dr. Kevin Brahm**  
**Key Account Manager International**

Phone +49 2064 427-280

Fax +49 2064 427-222

Email [k.brahm@ymc.eu](mailto:k.brahm@ymc.eu)

**Dr. Mathias Hehn**  
**Head of Laboratories**

Phone +49 2064 427-283

Telefax +49 2064 427-222

Email [m.hehn@ymc.eu](mailto:m.hehn@ymc.eu)

**Dr. Daniel Eßer**  
**Product Manager Analytical Chromatography**

Phone +49 2064 427-280

Fax +49 2064 427-222

Email [d.esser@ymc.eu](mailto:d.esser@ymc.eu)

**Ann Marie Rojahn**  
**Product Specialist Analytical Chromatography**

Phone +49 2064 427-280

Fax +49 2064 427-222

Email [a.rojahn@ymc.eu](mailto:a.rojahn@ymc.eu)

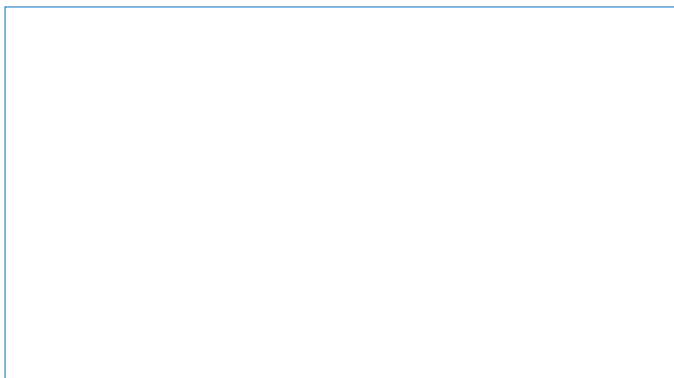
**Dr. Kirstin Arend**  
**Product Specialist Analytical Chromatography**

Phone +49 2064 427-280

Fax +49 2064 427-222

Email [k.arend@ymc.eu](mailto:k.arend@ymc.eu)

Your local distributor:



**YMC Europe GmbH**

Schöttmannshof 19  
D-46539 Dinslaken  
Germany  
Tel +49 2064 427-0, Fax +49 2064 427-222  
[www.ymc.eu](http://www.ymc.eu)

**YMC Schweiz GmbH**

Im Wasenboden 8  
4056 Basel  
Switzerland  
Tel +41 61561 80-50, Fax +41 61561 80-59  
[www.ymc-schweiz.ch](http://www.ymc-schweiz.ch)

**YMC CO., LTD.**

YMC Karasuma-Gojo Bld. 284 Daigo-cho,  
Karasuma Nishiiru Gojo-dori Shimogyo-ku,  
Kyoto 600-8106 Japan  
Tel +81 75342 45-15, Fax +81 75342 45-50  
[www.ymc.co.jp](http://www.ymc.co.jp)