

## Separation of methoxybenzenes

### If you want to go fast – go with YMC-Triart!

As a  $\pi$ -electron acceptor YMC-Triart Phenyl is perfectly suited for the separation of the aromatic methoxybenzenes. The U(H)PLC method also enables baseline separation of the three dimethoxybenzene and two trimethoxybenzene positional isomers.

### Less than 3 minutes for 5 methoxybenzenes on YMC-Triart Phenyl\*

\*stable under 100% aqueous conditions

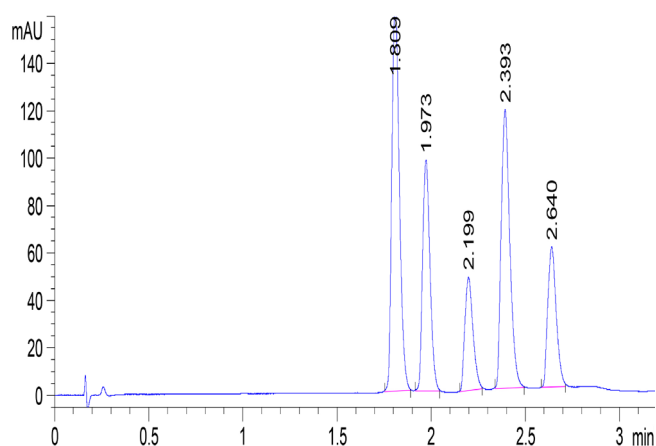


Figure 1: Separation of methoxybenzenes using YMC-Triart Phenyl.

Table 1: Method details

<b>Column</b>	YMC-Triart Phenyl 1.9 $\mu$ m, 12 nm, 50 x 2 mm ID	
<b>Part No.</b>	TPH12SP9-0502PT	
<b>Eluent</b>	A: Water B: Methanol	
<b>Gradient</b>	Time [min]	Eluent B [%]
	0	20
	4	80
	5	80
<b>Flow rate</b>	0.9 mL/min	
<b>Temperature</b>	25°C	
<b>Detection</b>	UV at 254 nm	
<b>Injection</b>	1.5 $\mu$ L dissolved in methanol	

Table 2: Chlorophenols analysed

Retention time	Analyte	Structure	Concentration
1.8 min	1,2-Dimethoxybenzene (Veratrole)		1 $\mu$ L/mL
1.9 min	1,2,4-Trimethoxybenzene		1 $\mu$ L/mL
2.1 min	1,4-Dimethoxybenzene		1 mg/mL
2.3 min	1,3-Dimethoxybenzene		1 $\mu$ L/mL
2.6 min	1,3,5-Trimethoxybenzene		1 mg/mL

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